

## Computer Simulations and Crossover Equation of State of Square-Well Fluids with Variable Width.

J.R. Elliott, Jr.,<sup>1,C,S</sup> S.B. Kiselev,<sup>2</sup> J.F. Ely,<sup>2</sup> and L. Lue<sup>2,3</sup>

<sup>1</sup>*Chemical Engineering Department  
University of Akron, Akron, OH 44325-3906, U.S.A.*

<sup>2</sup>*Chemical Engineering Department, Colorado School of Mines,  
Golden, Colorado 80401-1887, U.S.A.*

<sup>3</sup>*Department of Chemical Engineering, UMIST  
Manchester M60 1QD, UK*

The square-well systems capture the essential features of real materials while remaining simple enough to treat using analytic and simulation methods. Because of its simplicity, the square-well fluid (SWF) has long served as a model system for understanding the behavior of real fluids. In the present work, we perform extensive new molecular dynamics (MD) and Monte Carlo (MC) simulations in the one-phase region for the SWF with well widths of  $\lambda = 1.25, 1.375, 1.5, 1.75, 1.90, 2.0, 2.1$ , and  $3.0$ . These data together with MD and MC data reported earlier by other authors, have been used to develop a crossover equation of state (CR EOS) for square-well fluids with varying well width. The CR EOS for SWF yields the exact second and third virial coefficients, and accurately reproduces first order (high-temperature) perturbation theory results. In addition, the crossover equation of state contains a Ginzburg number,  $Gi$ , as a parameter, and asymptotically close to the critical point at  $|\tau| \ll Gi$  yields the correct scaling exponents for the coexistence curve  $\Delta\rho \propto |\tau|^\beta$ , isothermal susceptibility  $\chi_T \propto |\tau|^{-\gamma}$ , and isochoric heat capacity  $C_V \propto |\tau|^{-\alpha}$ . We fit the parameters of our equation of state to one-phase and two-phase thermodynamic data from our simulations and those of previous researchers. The resulting equation of state is found to represent the thermodynamic properties of these square-well fluids to less than 1% deviation in internal energy and density and 0.1% deviation in vapor pressure.